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Identification of unknown compounds is an essential skill for a chemist. Through observation, experiments and analysis, chemists can make a chemical logical judgment on unknowns.

Click the link: <http://www.tianenkeji.com/6matrix/index.html>, a 6×6 unknown set will be presented as below. The unknowns are identified by the letters A through L. Clicking on the buttons  allows you to observe reactions between the unknowns. Clicking on a letter around the edge will give a picture of the pure solution.

	G	H	I	J	K	L
A						
B						
C						
D						
E						
F						

The unknowns A ~ F are solutions of heavy metal compounds which are among reagents 1~10 given here:

No.	1	2	3	4	5
Compound	AgNO ₃	BaCl ₂	Cd(NO ₃) ₂	FeCl ₃	FeSO ₄
Conc. (M)	0.5	0.2	0.2	0.2	0.2
No.	6	7	8	9	10
Compound	Hg(NO ₃) ₂	MnCl ₂	Ni(NO ₃) ₂	Pb(NO ₃) ₂	Zn(NO ₃) ₂
Conc. (M)	0.2	0.2	0.2	0.2	0.2

The unknowns G ~ L are colorless and transparent solutions which come from the reagents 11~ 18:

No.	11	12	13	14
Compound	NaCl	NaBr	Na ₂ CO ₃	Na ₂ SO ₄
Conc. (M)	0.2	0.2	0.2	0.5
No.	15	16	17	18
Compound	NaOH	Na ₂ S	K ₂ C ₂ O ₄	NH ₃
Conc. (M)	0.2	0.2	0.5	2

Identify the compounds in the unknown matrix. Fill your determination on the answer sheet and write a balanced chemical reaction equation that is specific for clearly identifying this unknown compound.